

# KAUST VIRTUAL RESEARCH CONFERENCE 2021

## Enabling CO<sub>2</sub> Geological Storage within a Low-Carbon Economy



### CO<sub>2</sub> storage and utilization techniques: An overview

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# PRESENTATION OUTLINES

## INTRODUCTION

## LITERATURE REVIEW:

### a) CO<sub>2</sub> STORAGE TECHNIQUES

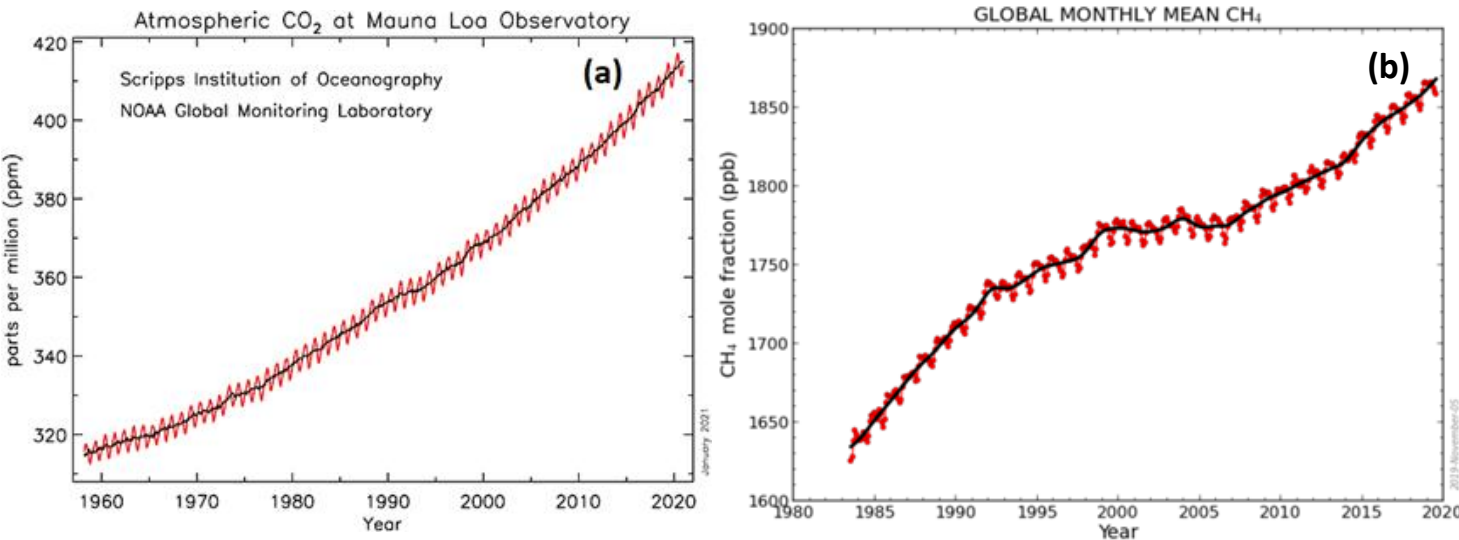
### b) CO<sub>2</sub> UTILIZATION TECHNIQUES

## CONCLUSIONS & OUTLOOK



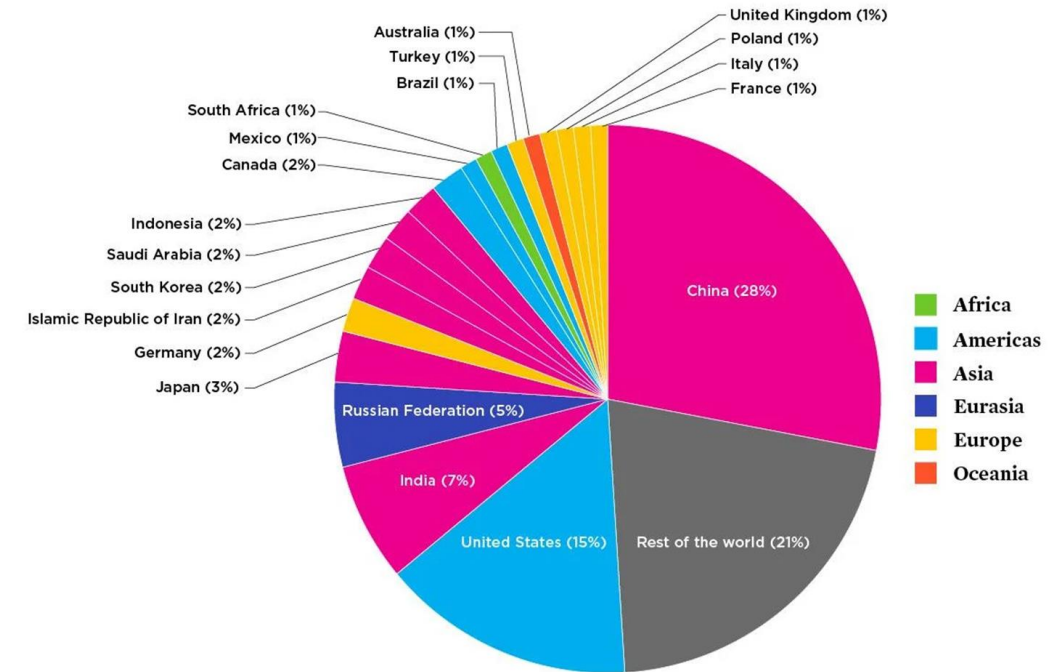
# INTRODUCTION

## World CO<sub>2</sub> & CH<sub>4</sub> Concentration Vs Time



**Figure 1:-** (a) Surface average atmospheric CO<sub>2</sub> concentration (ppm) [1], (b) Globally-averaged, monthly mean atmospheric CH<sub>4</sub> abundance [2], ESRL Global Monitoring Division, USA.

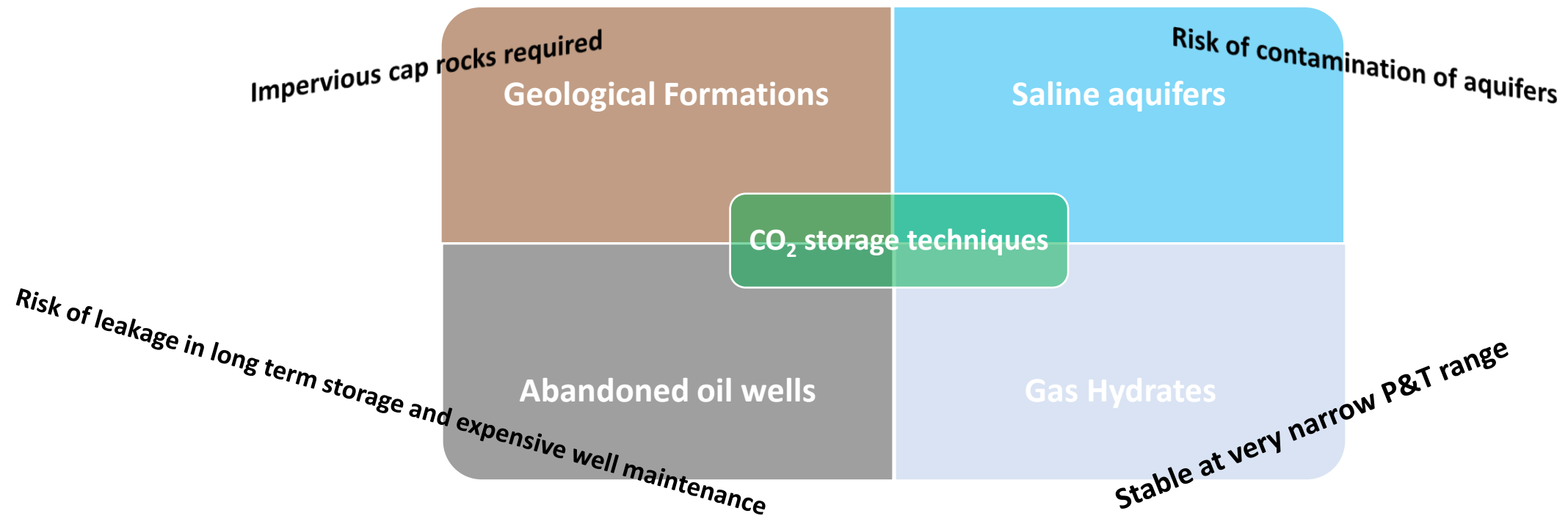
## CO<sub>2</sub> emission share from fossil fuels of different countries



**Figure 2:-** Pictorial view of CO<sub>2</sub> emission share from fossil fuels of different countries [3].



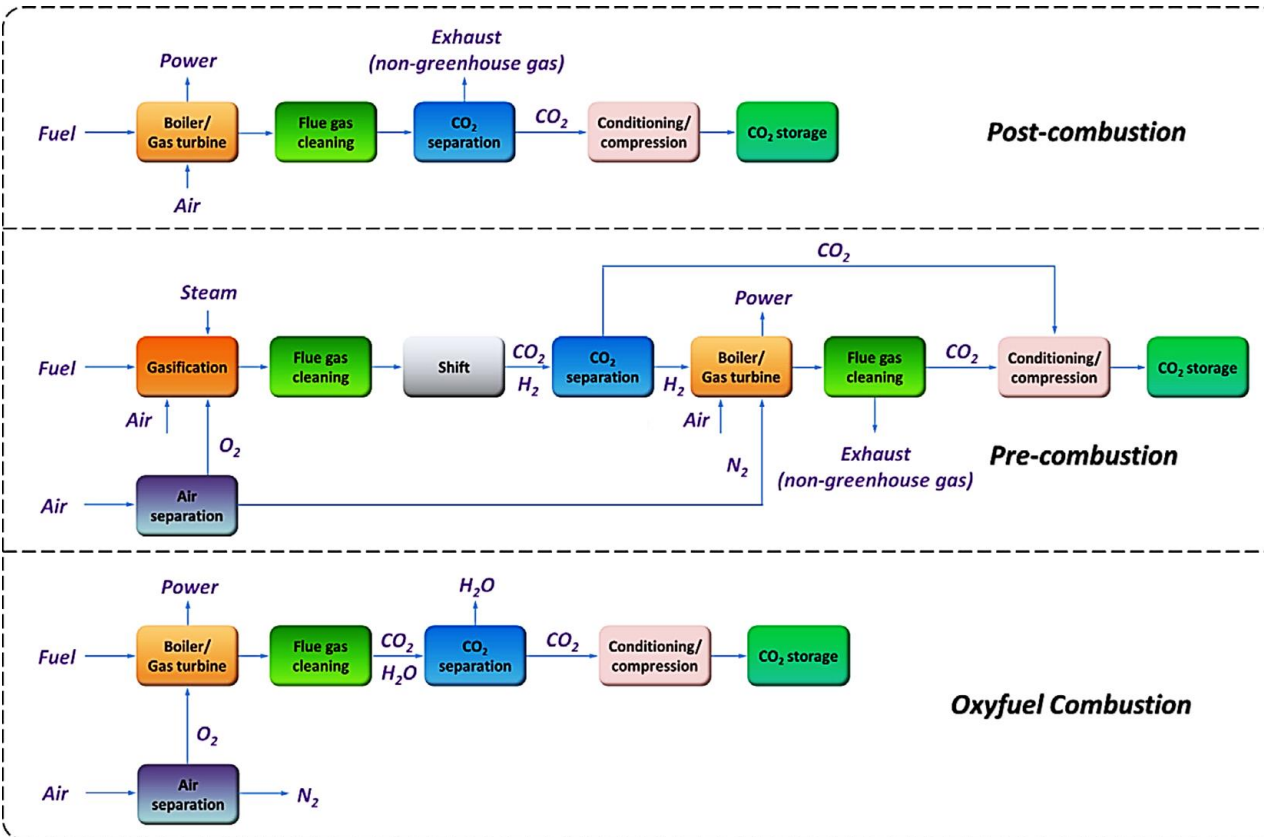
# CO<sub>2</sub> STORAGE TECHNIQUES



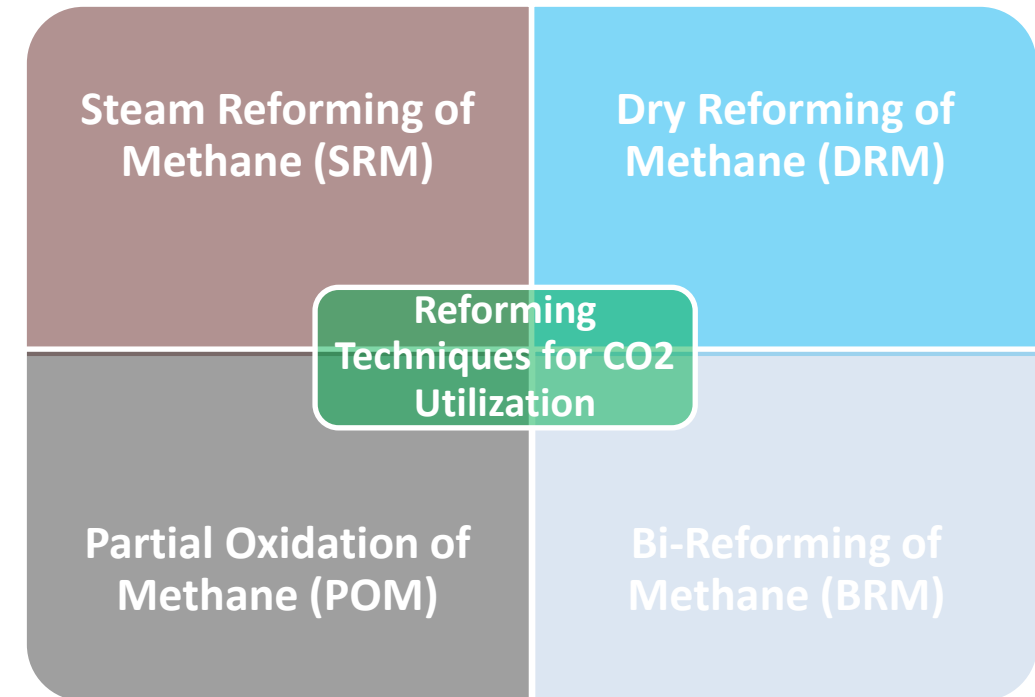
**Figure 3:** A pictorial view of the global greenhouse gas emissions [4,5]



# CO<sub>2</sub> UTILIZATIONS TECHNIQUES



**Figure 4:** A process flow diagram of CO<sub>2</sub> capture techniques [6]



**Figure 5:** Reforming techniques for CO<sub>2</sub> utilization [7-9]

# CONCLUSIONS & OUTLOOK

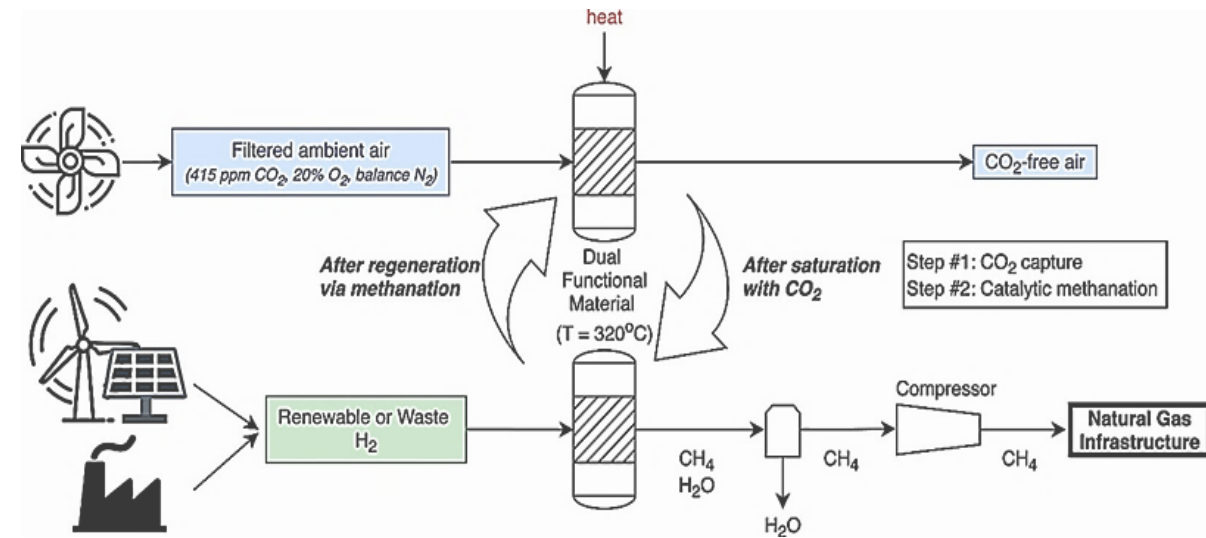
## CONCLUSION

Global warming is escalating due to tremendous amount of GHG's released into the atmosphere.

Numerous CO<sub>2</sub> storage techniques are available, but either lacks long term stable storage or uneconomic.

Several CO<sub>2</sub> utilization are being extensively studied but have limitation either in terms of commercialization or in its infancy stage. Hence continuous research is required for economic and technical barriers.

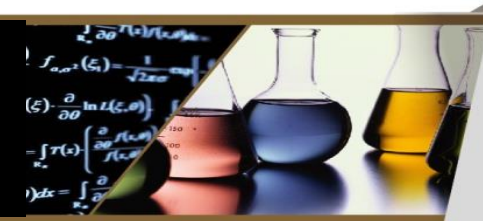
## OUTLOOK: Direct air capture (DAC) technique for CO<sub>2</sub> capture & utilization



**Figure 6:** Block diagram of DAC technique for CO<sub>2</sub> capture and storage [10]



# REFERENCES



- [1] E. Global, M. Division, G. Greenhouse, and G. Reference, "Global Monitoring Division (/ gmd /) Monthly Average Mauna Loa CO<sub>2</sub>," pp. 2–5, 2020.
- [2] E. Global, M. Division, G. Greenhouse, and G. Reference, "Global Monitoring Division (/ gmd /) Trends in Atmospheric Methane Global CH<sub>4</sub> Monthly Means," no. September 2018, pp. 1–6, 2020
- [3] 2020 Union of concerned scientist's data: Earth systems science data 11, 1783-1838, 2019.
- [4] M. Yusuf, NATURAL GAS HYDRATES : THE FUTURE'S FUEL, J. Environ. Res. Dev. 10 (2016) 738–746.
- [5] E. Dendy Sloan, Gas Hydrates: Review of Physical/Chemical Properties, Energy Fuels 1998, 12, 2, 191–196.
- [6] C. Song et al. Cryogenic-based CO<sub>2</sub> capture technologies: State-of-the-art developments and current challenges, Renew. Sustain. Energy Rev. 101 (2019) 265–278.
- [7] Yusuf M, Farooqi AS, Keong LK, Hellgardt K, Abdullah B. Contemporary trends in composite Ni-based catalysts for CO<sub>2</sub> reforming of methane. Chem Eng Sci
- [8] Yusuf Mohammad, Farooqi Ahmad Salam, Keoung Lau Kok, Hellgardt Klaus AB. Latest trends in Syngas production employing compound catalysts for methane dry reforming. I O P Conf Ser Mater Sci 2020.
- [9] A. Salam, M. Yusuf, ScienceDirect A comprehensive review on improving the production of rich-hydrogen via combined steam and CO<sub>2</sub> reforming of methane over Ni-based catalysts, Int. J. Hydrogen Energy. (2021).
- [10] C. Jeong-Potteret al. Feasibility Study of Combining Direct Air Capture of CO<sub>2</sub> and Methanation at Isothermal Conditions with Dual Function Materials, Appl. Catal. B Environ. 282 (2021) 119416.

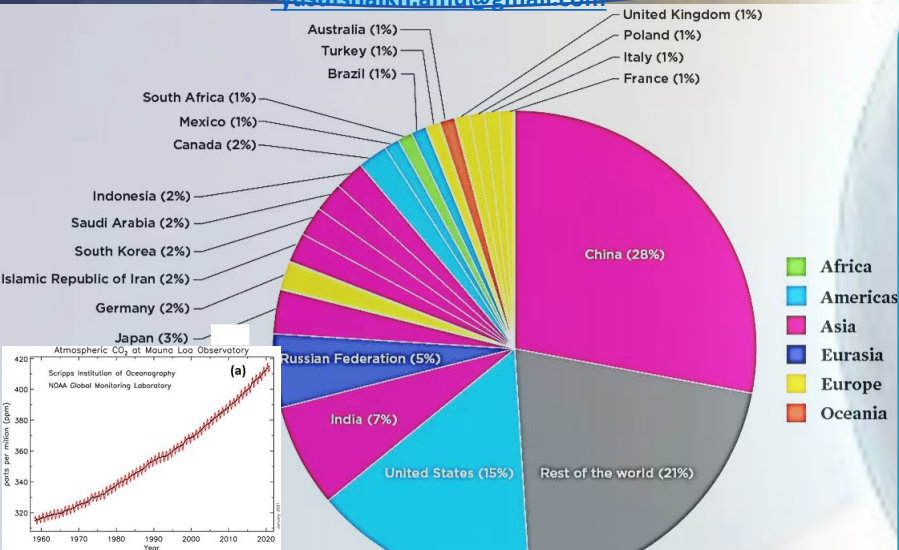


# CO<sub>2</sub> capture and utilization techniques: An overview

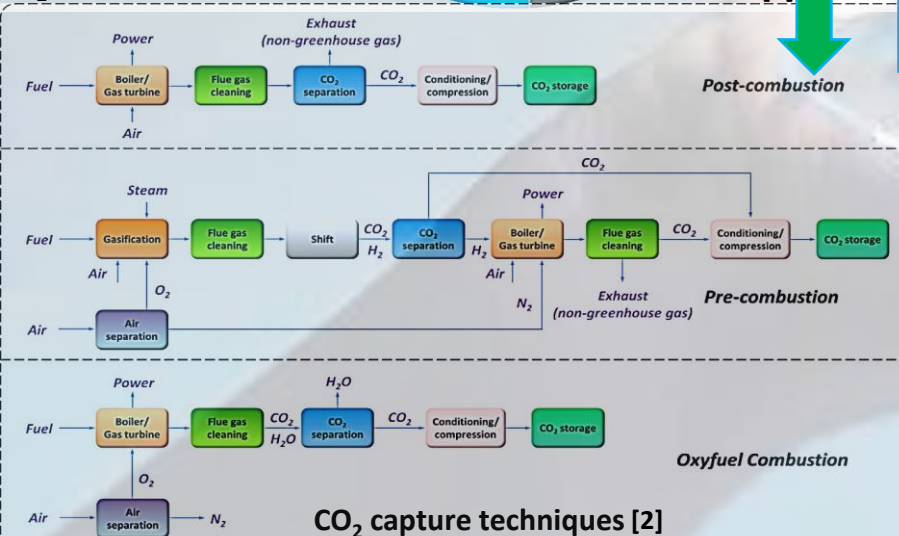
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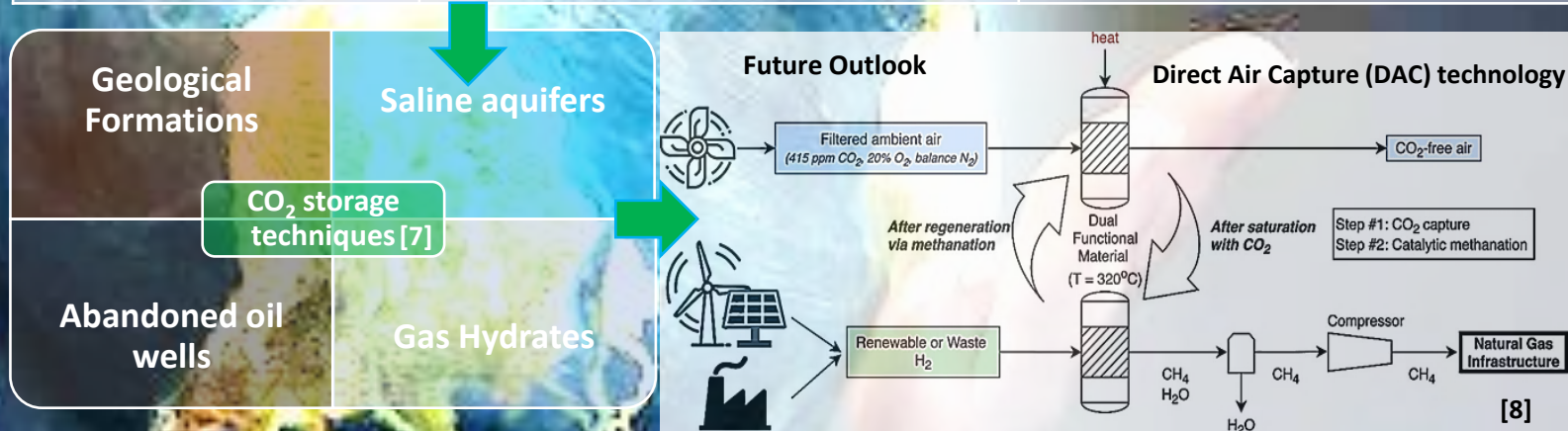


CO<sub>2</sub> emission share from fossil fuels of different countries [1]



CO<sub>2</sub> capture techniques [2]

CO <sub>2</sub> Capture Technique	Advantages	Limitations
Pre-combustion	<ul style="list-style-type: none"> <li>High CO<sub>2</sub> concentration (~45 vol%) and pressure.</li> <li>Commercial applied in some industrial sectors.</li> </ul>	<ul style="list-style-type: none"> <li>Energy penalty due to sorbent regeneration.</li> <li>Severe operating conditions (15–20 bar and 190–210 °C). [3,4]</li> </ul>
Oxy-fuel combustion	<ul style="list-style-type: none"> <li>Lower capital cost.</li> <li>High CO<sub>2</sub> concentration (80–98%).</li> <li>Low investment of boiler and other equipment.</li> </ul>	<ul style="list-style-type: none"> <li>High efficiency drop and energy penalty due to ASU. [5,6]</li> </ul>
Post-combustion	<ul style="list-style-type: none"> <li>A straightforward approach to be retrofitted</li> <li>More mature than other strategies</li> </ul>	<ul style="list-style-type: none"> <li>Dilute CO<sub>2</sub> concentration (5–15 vol%) at near atmospheric pressure.</li> <li>Energy penalty due to solvent/sorbent regeneration. [5]</li> </ul>



## References

- [1] 2020 Union of concerned scientist's data: Earth systems science data 11, 1783–1838, 2019.
- [2] C. Song et al. Cryogenic-based CO<sub>2</sub> capture technologies: State-of-the-art developments and current challenges, *Renew. Sustain. Energy Rev.* 101 (2019) 265–278.
- [3] Z. Dai et al. Recent advances in multi-layer composite polymeric membranes for CO<sub>2</sub> separation: A review, *Green Energy Environ.* 1 (2016) 102–128.
- [4] S. Nandi et al. A single-ligand ultra-microporous MOF for pre-combustion CO<sub>2</sub> capture and hydrogen purification, *Sci. Adv.* 1 (2015) 1–10.
- [5] N. Hedin et al. Adsorbents for the post-combustion capture of CO<sub>2</sub> using rapid temperature swing or vacuum swing adsorption, *Appl. Energy.* 104 (2013) 418–433.
- [6] J. Tonzello et al. Oxygen production technologies for IGCC power plants with CO<sub>2</sub> capture, *Energy Procedia.* 4 (2011) 637–644.
- [7] M. Yusuf, NATURAL GAS HYDRATES : THE FUTURE'S FUEL, *J. Environ. Res. Dev.* 10 (2016) 738–746.
- [8] C. Jeong-Potteret al. Feasibility Study of Combining Direct Air Capture of CO<sub>2</sub> and Methanation at Isothermal Conditions with Dual Function Materials, *Appl. Catal. B Environ.* 282 (2021) 119416.